Fluorescence Efficiency of Surface Seawater as a Function of Excitation and Emission Wavelength

by Sarah A. Green

Fluorescence of natural waters has often been employed in attempts to quantify dissolved chromophores. However, it has not always been recognized that the intensity of fluorescence obtained from a given water sample depends on the fluorescence efficiency of the absorbing components as well as on the concentration of light-absorbing material present. Although fluorescence intensities have been compared for a variety of seawater samples (Willey and Atkinson, 1982; Hayase et al., 1988; Chen and Bada, 1989), there have been few measurements of the efficiency of emission in natural waters (Zepp and Schlotzhauer, 1981; Ferrari and Tassan, 1991). As one part of my Ph.D. work, I determined quantum efficiencies as a function of excitation wavelength for a series of surface-seawater samples.

Fluorescence quantum yield ($\phi$) is defined as the ratio of emitted to absorbed photons. For any natural water, $\phi$ depends both on the types of chromophores present and on their relative concentrations. Yet, it is always independent of dilution factors. In contrast to a system containing only a single chromophore, in a mixture $\phi$ may be a function of excitation wavelength. Determination of $\phi$ requires accurate measurement of absorption coefficients and emission intensities over the UV-visible range. Because isolation techniques for organic carbon can change the distribution of chromophores in a sample, (Green, 1992) it is preferable to measure fluorescence on water that is unaltered except for removal of particles by filtration. However, this proves difficult in very clear oceanic waters where absorption of unconcentrated samples is below the detection limits of available instruments; fluorescence is still observable in these waters because of the inherently greater sensitivity of the technique.

In order to obtain a full spectral map of fluorescence efficiency, I combined absorption data with excitation/emission matrix plots that provide a map of fluorescence intensity over a range of wavelengths (Coble et al., 1990) (Fig. 1). Intensities at each excitation wavelength have been divided by the absorption at that wavelength; thus each point of this three-dimensional graph represents the fraction of photons emitted at a particular frequency (right axis) per photon absorbed at the corresponding excitation wavelength indicated on the left axis. Integration of the emission spectra gives a plot of quantum yield versus excitation wavelength (left); integration of excitation spectra gives the total emission obtained under broad-band light (equal intensity 260–470 nm). A quinine sulfate solution (OD = 0.1, in 1 N H$_2$SO$_4$) was used to calibrate the fluorometer output to quantum yield.

Fluorescence intensity maxima of natural organic matter are generally observed at excitation and emission wavelengths of 345 and 445 nm, respectively, with an additional band appearing for short-wavelength excitation (~ 300 nm) (Coble et al., 1990). In contrast, Figure 1 shows that maximum fluorescence efficiency is obtained for 395 nm excitation, with emission centered at 480 nm. The decrease in efficiency for excitation below 350 nm demonstrates that, although light absorption increases nearly exponentially with shorter wavelengths, emission does not increase in proportion.

The shape of fluorescence efficiency plots was surprisingly consistent for surface waters collected in the Gulf of Mexico, Oyster Bay (Everglades National Park), the Amazon and Orinoco Rivers, and the Caribbean Sea, as well as for dissolved organic carbon (DOC) isolated from the Sargasso Sea at depths of 50–3,200 m. In addition, the quantum efficiency of fluorescence at a reference excitation wavelength ($\lambda_{ex} = 355$ nm) fell in a narrow range of 0.75–2% for all nat-
ural samples examined. This suggests that fluorescence, at defined excitation and emission points, is a reliable indicator of absorbing material and encourages current attempts to employ fluorescence detection for quantification of dissolved chromophores by remote sensing methods (Hoge and Swift, 1982, 1985; Bristow et al., 1985). In current work, fluorescence efficiency plots are being combined with the solar spectrum to predict the total emission spectrum of solar-induced fluorescence from surface waters of varying organic content (Vodacek, 1992). This information can then be employed to increase the accuracy of algorithms used in the estimation of phytoplankton populations from satellite observations.

Three-dimensional fluorescence efficiency matrices provide essential information for the development of photon budgets for the oceans (Smith et al., 1989). Further work in this area should focus on more accurate methods for measuring absorbance in very weakly absorbing seawater samples so that quantum yields can be measured on unconcentrated deep-sea DOC (evidence from isolated DOC from the Sargasso Sea) suggests that subsurface samples may have higher fluorescence quantum yields than surface waters). Measurement of absorbance and fluorescence during and immediately after plankton blooms also should be pursued.

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References


